

Operating Instructions

ISEmax CAS40D

Ion-selective sensor for continuous measurement of ammonium, nitrate and other ions

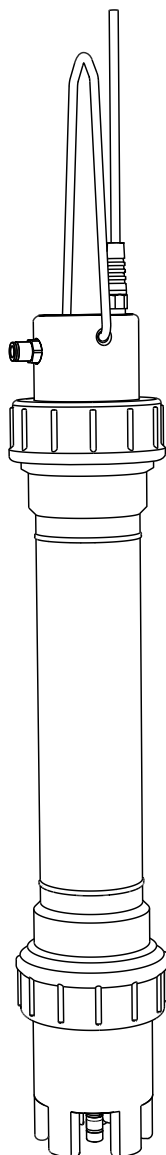





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1 About this document

1.1 Warnings

Structure of information	Meaning
 DANGER Causes (/consequences) If necessary, Consequences of non-compliance (if applicable) ► Corrective action	This symbol alerts you to a dangerous situation. Failure to avoid the dangerous situation will result in a fatal or serious injury.
 WARNING Causes (/consequences) If necessary, Consequences of non-compliance (if applicable) ► Corrective action	This symbol alerts you to a dangerous situation. Failure to avoid the dangerous situation can result in a fatal or serious injury.
 CAUTION Causes (/consequences) If necessary, Consequences of non-compliance (if applicable) ► Corrective action	This symbol alerts you to a dangerous situation. Failure to avoid this situation can result in minor or more serious injuries.
NOTICE Cause/situation If necessary, Consequences of non-compliance (if applicable) ► Action/note	This symbol alerts you to situations which may result in damage to property.


1.2 Symbols

Symbol	Meaning
	Additional information, tips
	Permitted or recommended
	Not permitted or not recommended
	Reference to device documentation
	Reference to page
	Reference to graphic
	Result of a step

2 Basic safety instructions

2.1 Requirements for personnel

- Installation, commissioning, operation and maintenance of the measuring system may be carried out only by specially trained technical personnel.
- The technical personnel must be authorized by the plant operator to carry out the specified activities.
- The electrical connection may be performed only by an electrical technician.
- The technical personnel must have read and understood these Operating Instructions and must follow the instructions contained therein.
- Faults at the measuring point may only be rectified by authorized and specially trained personnel.

 Repairs not described in the Operating Instructions provided must be carried out only directly at the manufacturer's site or by the service organization.

2.2 Designated use

The ion-selective sensor is designed for measuring tasks in the activated sludge basin and in the inlet to the activated sludge basin of municipal sewage treatment plants.

The following parameters can be monitored and regulated depending on the device version:

- Nitrate
- Ammonium
- Potassium (also to compensate ammonium)
- Chloride (also to compensate nitrate)
- pH value
- ORP

Use of the device for any purpose other than that described, poses a threat to the safety of people and of the entire measuring system and is therefore not permitted.

The manufacturer is not liable for damage caused by improper or non-designated use.

2.3 Workplace safety

As the user, you are responsible for complying with the following safety conditions:

- Installation guidelines
- Local standards and regulations

Electromagnetic compatibility

- The product has been tested for electromagnetic compatibility in accordance with the applicable international standards for industrial applications.
- The electromagnetic compatibility indicated applies only to a product that has been connected in accordance with these Operating Instructions.

2.4 Operational safety

Before commissioning the entire measuring point:

1. Verify that all connections are correct.
2. Ensure that electrical cables and hose connections are undamaged.
3. Do not operate damaged products, and protect them against unintentional operation.
4. Label damaged products as defective.

During operation:

- ▶ If faults cannot be rectified:
products must be taken out of service and protected against unintentional operation.

CAUTION

Cleaning not switched off during calibration or maintenance activities

Risk of injury due to medium or cleaning agent!

- ▶ If a cleaning system is connected, switch it off before removing a sensor from the medium.
- ▶ If you wish to check the cleaning function and have therefore not switched off the cleaning system, wear protective clothing, goggles and gloves or take other appropriate measures.

2.5 Product safety

The product is designed to meet state-of-the-art safety requirements, has been tested, and left the factory in a condition in which it is safe to operate. The relevant regulations and international standards have been observed.

3 Incoming acceptance and product identification

3.1 Incoming acceptance

1. Verify that the packaging is undamaged.
 - ↳ Notify the supplier of any damage to the packaging.
Keep the damaged packaging until the issue has been resolved.
2. Verify that the contents are undamaged.
 - ↳ Notify the supplier of any damage to the delivery contents.
Keep the damaged goods until the issue has been resolved.
3. Check that the delivery is complete and nothing is missing.
 - ↳ Compare the shipping documents with your order.
4. Pack the product for storage and transportation in such a way that it is protected against impact and moisture.
 - ↳ The original packaging offers the best protection.
Make sure to comply with the permitted ambient conditions.

If you have any questions, please contact your supplier or your local Sales Center.

3.2 Product identification

3.2.1 Nameplate

The nameplate provides you with the following information on your device:

- Manufacturer identification
 - Order code
 - Extended order code
 - Serial number
 - Ambient and process conditions
 - Input and output values
 - Safety information and warnings
- Compare the information on the nameplate with the order.

3.2.2 Product identification

Product page

www.endress.com/cas40d

Interpreting the order code

The order code and serial number of your product can be found in the following locations:

- On the nameplate
- In the delivery papers

Obtaining information on the product

1. Go to www.endress.com.
2. Call up the site search (magnifying glass).
3. Enter a valid serial number.
4. Search.
 - ↳ The product structure is displayed in a popup window.

5. Click on the product image in the popup window.
 - ↳ A new window (**Device Viewer**) opens. All of the information relating to your device is displayed in this window as well as the product documentation.

3.3 Scope of delivery

The scope of delivery comprises:

- 1 sensor, version as ordered
- 1 socket wrench
- 1 tube silicone grease
- 1 Operating Instructions

3.4 Certificates and approvals

3.4.1 CE mark

Declaration of Conformity

The product meets the requirements of the harmonized European standards. As such, it complies with the legal specifications of the EU directives. The manufacturer confirms successful testing of the product by affixing to it the **CE** mark.

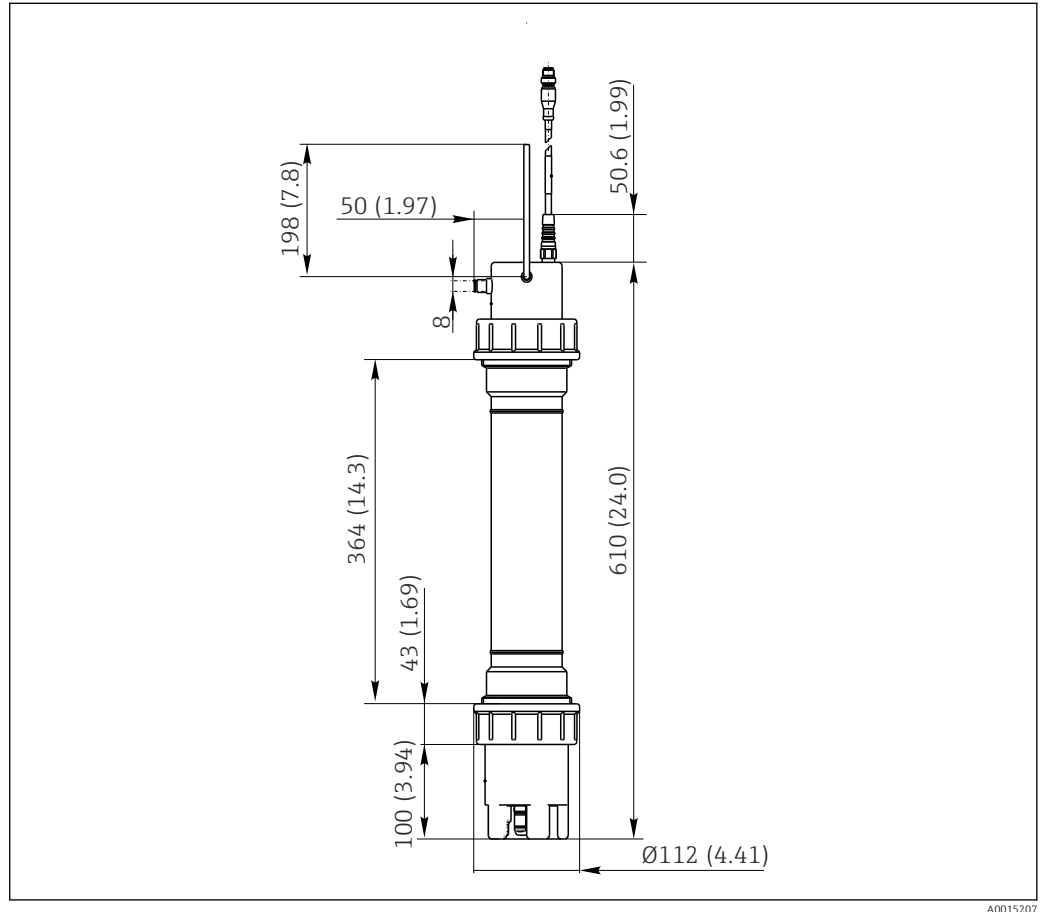
3.4.2 EAC

The product has been certified according to guidelines TP TC 004/2011 and TP TC 020/2011 which apply in the European Economic Area (EEA). The EAC conformity mark is affixed to the product.

4 Installation

4.1 Installation conditions

4.1.1 Dimensions



1 Dimensions in mm (inch)

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4.1.2 Mounting location

Choose a mounting location that can be easily accessed at a later stage.

- Ensure that upright posts and assemblies are fully secured and vibration-free.

4.2 Mounting the sensor

4.2.1 Electrode installation

NOTICE

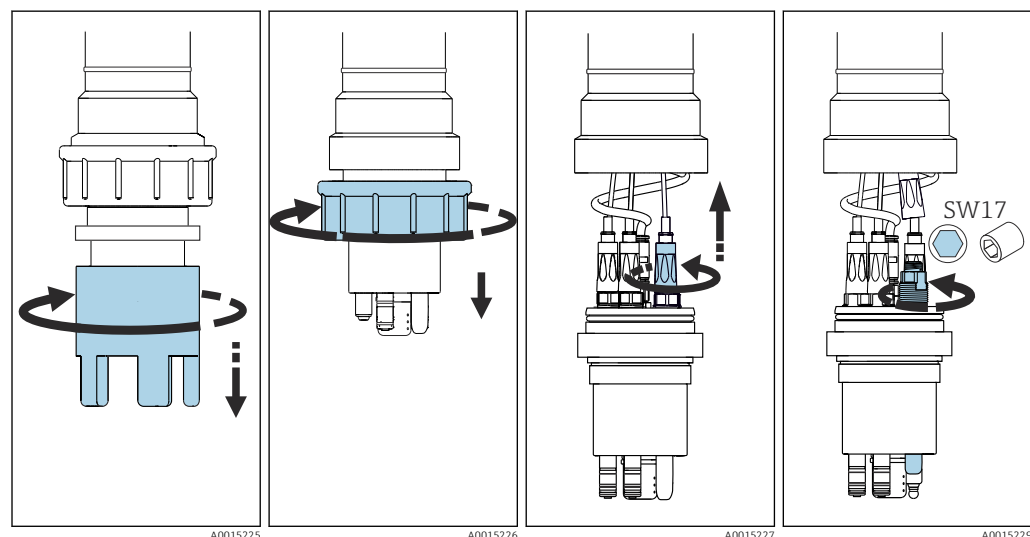
Electrode protection caps not used or incorrectly used

Drying out of pH electrode or damage to ion-selective membrane

- ▶ Remove the protection cap from the pH electrode before immersing the sensor in the medium.
- ▶ Set the protection cap aside.
- ▶ If you remove the sensor from the medium for more than 20 minutes, place the protection cap filled with 1-3 M KCl solution back on the pH electrode. This prevents the electrode from drying out.
- ▶ pH electrodes that have dried out due to incorrect storage can be made ready for measurement again by leaving them in 3 M KCl solution for up to 12 hours.
- ▶ The ion-selective electrodes do not have a protection cap. Never put one of these caps on the electrodes.

i All electrodes are installed and wired in the factory in accordance with the version ordered.

Installing an additional electrode (optional)



- 2 Release protection guard
 3 Remove coupling nut
 4 Release cable
 5 Remove electrode

1. Release the protection guard and remove (→ 2, 10).
2. Unscrew the coupling nut (→ 3, 10).
3. Remove the electrode holder from the sensor.
4. Release the electrode cable on a dummy electrode (placeholder, must be present to ensure tight sealing → 4, 10).
5. Remove the dummy electrode using a socket wrench, AF17 (→ 5, 10).
6. Install the new electrode in the free space.
7. Tighten the electrode hand-tight with the socket wrench, AF17.
8. Connect the electrode connector.
9. Pay attention to the color coding of the electrodes and the labeling of the cable. Refer to the following table for this purpose → 11.
10. Carefully push the electrode holder and the air hose back into the sensor.
11. Screw on the coupling nut followed by the protection guard.

NOTICE**Air bubbles**

Air bubbles may accumulate in the electrodes following transportation and if they have been stored in a horizontal position. These air bubbles cause measuring errors.

- ▶ Before installing the sensor, eliminate any air bubbles e.g. by shaking gently.
- ▶ Then make sure to hold the sensor in an upright position (electrodes facing downwards) at all times until the sensor is installed at the measuring point.


Electrode identification

Electrode	Color of membrane ring and marking on screw head ¹⁾	Cable identification
Ammonium	RD	1, 2 or 3
Nitrate	BU	
Potassium	YE	
Chloride	GN	
pH (incl. reference)	No marking	R
Temperature	No marking	T

1) color codes as per IEC 757

4.2.2 Installing at the measuring point**NOTICE****Compressed air**

Damage to the relay!

- ▶ The compressed air supply must not exceed 3.5 bar (50 psi).
- ▶ The compressed air must be supplied through an air filter (5 µm). This filter is already installed in the optional cleaning unit →  29.

Installing at the measuring point

1. Install additional electrodes in the sensor if necessary.
2. Connect the electrodes to the suitable cable connector.

3. NOTICE**Sensor too deep in the medium, tension in sensor cable.**

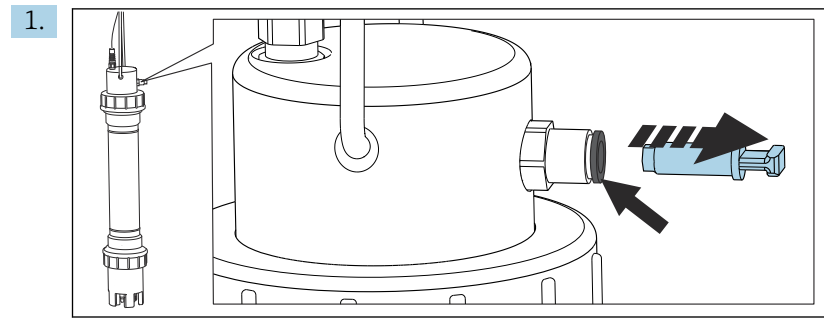
Failure of sensor due to penetration of medium or damage to cable!

- ▶ Do not use the cable to suspend the sensor into the medium. Use a suitable holder.
- ▶ Never use the cable to pull the sensor out of the medium.
- ▶ Never immerse the sensor completely in the medium.

Suspend the sensor from the chain on the holder.

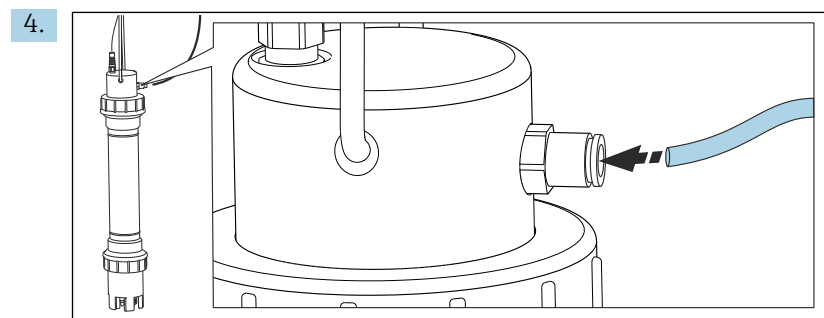
4. Adjust the chain length and the cross bearer for the holder in such a way that the sensor is immersed approx. 0.5 m (1.64 ft) in the medium and is approx. 0.5 m (1.64 ft) from the basin rim.
5. Route the cables in such a way that no mechanical damage or interference influences can arise from other cables.
6. Connect the optional cleaning unit to the transmitter and the pressure hose (OD 8) to the sensor.

Connecting an optional cleaning unit or an external compressed air supply



Remove the watertight dummy plug from the sensor's compressed air connection.

2. Push against the black ring for this purpose.
 3. Remove the plastic dummy plug.



Plug the compressed air hose (OD 8) belonging to the cleaning unit or compressed air supply into the compressed air connection.

5. Only optional cleaning unit:
 Connect the cleaning unit to the transmitter (for more information, see the Operating Instructions for the transmitter).

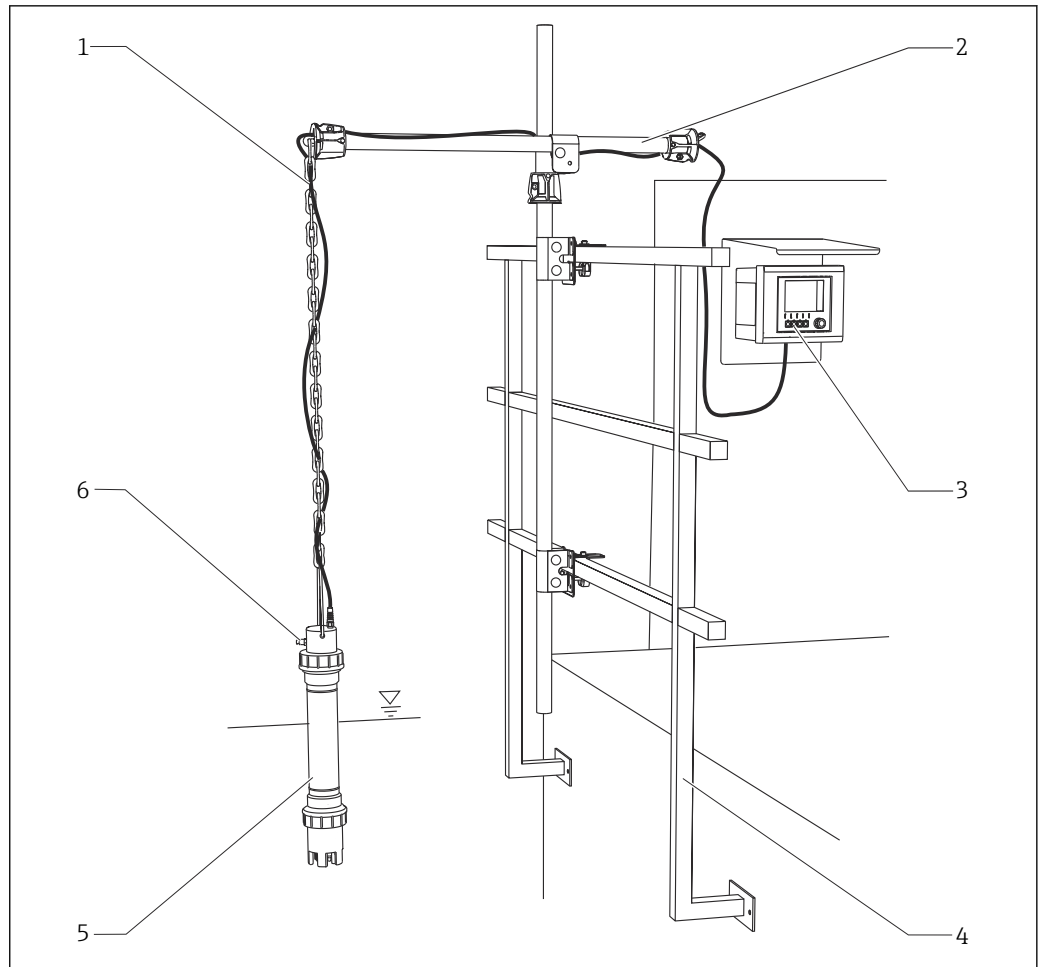
4.3 Installation example

A complete measuring system comprises:

- Sensor CAS40D
 - Ion-selective electrode(s) for ammonium, nitrate, potassium or chloride
 - pH glass electrode, Orbisint CPS11-1AS2GSA
 - Temperature sensor, CTS1
- Transmitter Liquiline CM44x

Optional:

- Assembly holder, e.g. CYH112
- Weather protection cover: absolutely essential if mounting the transmitter outdoors!
- Compressed air generator (if no compressed air available on site)



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6 Example: measuring system on basin rim

- 1 Sensor cable
- 2 Wastewater assembly holder, secure to rail, with transverse pipe and chain
- 3 Liquiline CM44x transmitter (in graphic: wall-mounted with weather protection cover)
- 4 Rail
- 5 Sensor CAS40D with ion-selective electrodes
- 6 Connection for optional compressed air cleaning (not in graphic)

4.4 Post-installation check

1. After mounting, check all the connections to ensure they are secure and leak-tight.
2. Check all cables and hoses for damage.
3. Verify that the cables are routed in such a way that they are free from electromagnetic interference influences.

5 Electrical connection

⚠ WARNING

Device is live!

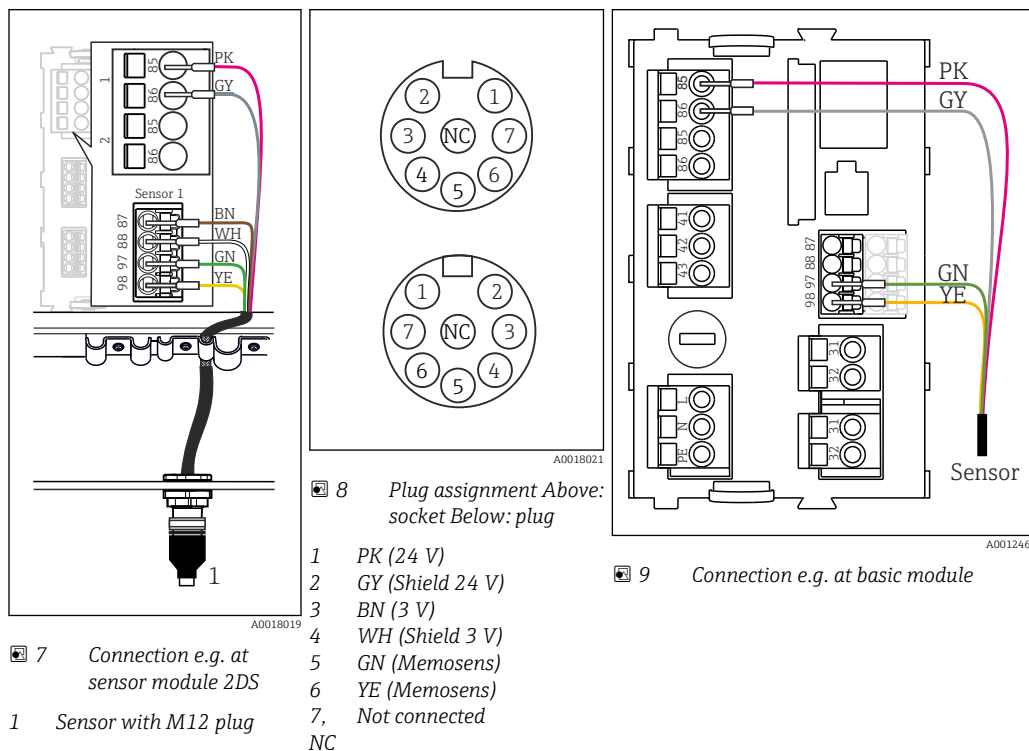
Incorrect connection may result in injury or death!

- ▶ The electrical connection may be performed only by an electrical technician.
- ▶ The electrical technician must have read and understood these Operating Instructions and must follow the instructions contained therein.
- ▶ **Prior** to commencing connection work, ensure that no voltage is present on any cable.

5.1 Connecting the sensor

When connecting to the Liquiline CM44x transmitter, there are two options:

1. M12 plug (version: fixed cable, M12 plug)
 - ↳ The wiring for the M12 socket is inside the device. Only the sensor plug is connected to the socket.
2. Direct connection of the fixed cable to the plug-in terminals (version: fixed cable, ferrules)



The maximum cable length is 100 m (328 ft).

5.2 Connecting additional electrodes in the sensor

All electrodes are connected in the factory prior to delivery.

Installing and connecting additional electrodes

- ▶ Install the electrode (→ 10).

Then reset the electrode configuration at the transmitter.

5.3 Ensuring the degree of protection

Only the mechanical and electrical connections which are described in these instructions and which are necessary for the required, designated use, may be carried out on the device delivered.

- Exercise care when carrying out the work.

Otherwise, the individual types of protection (Ingress Protection (IP), electrical safety, EMC interference immunity) agreed for this product can no longer be guaranteed due, for example to covers being left off or cable (ends) that are loose or insufficiently secured.

5.4 Post-connection check

Device status and specifications	Notes
Are the sensor and cable free from damage on the outside?	Visual inspection

Electrical connection	Notes
Does the supply voltage of the connected transmitter match the data on the nameplate?	Visual inspection
Are the installed cables strain-relieved and not twisted?	
Is the cable type route completely isolated on site?	Power cables / signal lines
Are all cable entries installed, tightened and sealed?	In the case of lateral cable entries: Cable loops facing downwards to allow water to drip off.
Are all cable entries facing downwards or mounted laterally?	

6 Commissioning

Select the correct pH electrode at the transmitter.

1. Path in the transmitter menu: **Setup/Inputs/ISE/1 (R) pH**
2. **Reference electrode:** Specify the pH electrode version, **Standard** or **Salt ring**.

The version of the pH electrode can only be found on the electrode's nameplate (CPS11-1AS*** = **Salt ring**, CPS11-1AT*** = **Standard**).

-  Sensors from 2019 onwards are always delivered with pH electrodes with salt storage (salt ring).

7 Operation

7.1 Adapting the measuring device to the process conditions

7.1.1 Calibration

Factory calibration

Before being delivered, the sensor is checked in the factory and precalibrated with regard to the sensor slope and zero point.

Since the correct calibration state depends on the medium matrix (ionic strength, concentration of interference ions etc.), users must always calibrate the sensor themselves after commissioning in order to adjust the zero point so that it suits the users' particular application conditions. The manual offset is set to zero upon delivery. If a compensation electrode is not used for automatic interference ion compensation, the offset must be set before the first calibration is performed if working with ammonium and nitrate electrodes.

Calibration recommendations

Application	Variables to be calibrated	Recommended calibration types
Commissioning	Zero point, manual offset	1-point calibration
Maintenance	Slope	Data entry Set the slope specified on the manufacturer's certificate in the transmitter
	Zero point	1-point calibration
Routine calibration	Zero point	1-point calibration

Types of calibration

- pH electrode:
 - 2-point calibration (recommended)
 - 1-point calibration
- Ion-selective electrodes:
 - 1-point calibration (recommended)
 - Data entry
 - 2-point calibration
 - Standard addition ("Expert" only)
- ORP sensor:
 - 1-point calibration
- Temperature adjustment by entering a reference value

Calibration parameters

When determining ion concentrations potentiometrically, the electrochemical measuring cell consists of the ion-selective electrode and a reference electrode. This cell supplies a voltage within the "linear" or preferably "NERNST" range which is proportional to the logarithm of the concentration (or activity) of the ions to be determined. The slope and zero point calibration parameters refer to this logarithmic relationship, which gives these parameters a completely different meaning in this measurement method compared with other measurement methods.

Slope

The slope is specified as a %, based on the theoretical slope according to Nernst.

Example: 98% slope = $59.16 \text{ mV/pX} \cdot 0.98 = 57.98 \text{ mV/pX}$

The slope affects the linearity of the measurement.

If the slope set at the transmitter is less than or greater than the actual slope of the ion-selective electrode, measured errors can occur due to non-linearities. The greater the concentration range in which the measured values vary, the greater the possible non-linearities. If, on the other hand, the measured values only vary in a small range, even larger slope errors will not result in discernible non-linearities. The slope is determined in the factory for every ion-selective electrode and every membrane cap and is indicated on the manufacturer's certificate supplied with the device. The user simply has to enter the slope data to communicate the supplied slope value to the transmitter. As the slope only changes marginally while the unit is in operation, the user does not normally have to perform a calibration. The slope is a property of the ion-selective electrode. Therefore the reference electrode does not affect the slope.

Slope of ion-selective electrodes

Electrode	Maximum	Minimum
Ammonium	110%	90%
Nitrate		90%, typically 98 - 100%
Potassium		90%
Chloride		

If the calibrated slope is outside the table values, the calibration conditions must be taken into account. Check if the manual offset or calibration of the compensation electrode is correct.

Zero point

The zero point determines the sensitivity of the measurement. If the configured zero point is too low or too high in relation to the actual zero point of the ion-selective electrode system, all the measured values are either too high or too low by a certain percentage. The zero point depends on the internal solution used by the ion-selective electrode and the reference electrode. As a result of the ion-selective electrode and the reference electrode aging, the zero point gradually changes over time and must be calibrated periodically. The zero point depends on both the ion-selective electrode and the reference electrode.

Typical zero points

Electrode	Typical zero point ¹⁾
Ammonium	1.1
Nitrate	1.4
Potassium	3.55
Chloride	-0.5

1) for new reference electrode (electrode aging affects the zero point)

Sequence for calibration/adjustment of measuring point

Some measured values from other electrodes or sensors are used for measured value compensation of ion-selective electrodes:

- Measured value of temperature sensor for temperature compensation
- pH measured value for pH compensation of ammonium (optional)
- Potassium or chloride measured value for compensation of interference ions in the case of ammonium or nitrate (optional)

For this reason, there is a sequence for calibration and adjustment that must be followed in order to achieve a reliable measurement:

1. Temperature adjustment (precalibrated in the factory, therefore not necessary for initial calibration)
2. Calibration and adjustment of pH electrode
3. Depending on whether compensation electrodes are used:
Calibration and adjustment of ion-selective compensation electrodes (potassium, chloride)
4. If no compensation electrodes are used:
A correct manual offset is configured for the ammonium and nitrate electrode
5. Calibration and adjustment of ion-selective measuring electrodes (ammonium, nitrate)

Calibration

The following minimum concentrations apply for 1-point and 2-point calibration:

- 6.4 mg/l ammonium or 5 mg/l ammonium-nitrogen
- 22.1 mg/l nitrate or 5 mg/l nitrate-nitrogen
- 20 mg/l potassium
- 100 mg/l chloride

The values are benchmarks which may change over time due to the influence of interference ions or the aging of ion-selective electrodes. If the calibration concentrations are too low, the measured values will be incorrect.

Stability criterion

The factory setting on the transmitter is "weak". The measured value of an ion-selective electrode only reaches adequate stability after approx. 4 min.

- Wait for a stable measuring signal before starting calibration.

Data entry

The zero point and the slope of the ion-selective electrode system can be entered directly and modified using the "Data entry" method.


When installing an ion-selective electrode or membrane cap in the sensor:

1. Set the electrode slope for the slot using "Data entry". The electrode slope is indicated on the manufacturer's certificate.
2. Calibrate the zero point.

1-point calibration

In the case of 1-point calibration, the zero point of the ion-selective electrode system is calibrated in a solution with a known concentration.

- Enter the reference value either before or after the measured value is recorded.
- Set the slope and the manual offset correctly or carry out a calibration of the compensation electrodes for ammonium and nitrate.


 Two ion-selective electrodes can be calibrated simultaneously (ammonium and nitrate or potassium and chloride) with the Liquiline CM44x transmitter.

1. Suspend the sensor into a vessel or into a process with a known concentration.
 - ↳ Experience shows that at 7 mg/l good values are produced during calibration for ammonium and nitrate.
2. Start the 1-point calibration in the transmitter menu.
 - ↳ Select whether or not the measured value of the reference medium is known.
3. Wait for the signal (mV value) to stabilize (approx. 4 min for new membrane caps).

4. Start calibration process.
 - ↳ Accept the calibration.

2-point calibration

In the case of 2-point calibration, the zero point and the slope of the ion-selective electrode system are determined using 2 solutions whose concentrations are known. The two concentrations in the two solutions should be in the upper and lower measuring range. When using 2-point calibration, the manual offset must already be set correctly, as non-linearities would otherwise not be corrected by the 2-point calibration.

 With 2-point calibration, the concentration should at least double. In this case, the change in the mV signal is approx. 1/3 of the slope in mV.

1. Suspend the sensor into a vessel or into a process with a known concentration.
 - ↳ Experience shows that at 7 mg/l good values are produced during calibration for ammonium and nitrate.
2. Start the 2-point calibration in the transmitter menu.
 - ↳ Select whether or not the measured value of the reference medium is known.
3. Wait for the signal (mV value) to stabilize (approx. 4 minutes for new membrane caps).
4. Start calibration process.
5. Clean sensor and quickly wipe it dry.
 - ↳ Suspend the sensor into the vessel with the second concentration.
6. Wait for the signal (mV value) to stabilize (approx. 4 minutes for new membrane caps).
7. Start calibration process.
 - ↳ Accept calibration.

Potassium and chloride compensation

Depending on the selectivity of the ion-selective electrode vis-à-vis other ions (interference ions), and the concentration of these ions, such ions could also be interpreted as part of the measuring signal and thus cause measuring errors. When measuring in wastewater, the potassium ion which is chemically similar to the ammonium ion can cause higher measured values. High concentrations of chloride may result in the measured values for nitrate being too high. To reduce measuring errors resulting from such cross-interference, the concentration of the potassium or chloride interference ion can be measured and compensated for with a suitable additional electrode. As an alternative to using compensation electrodes, a manual offset can be entered.

If compensation electrodes are used, there is no need to set a manual offset.

- Use of the potassium electrode for compensation:
For concentrations > 40 mg/l (> 40 ppm) with simultaneously fluctuating values of ± 20 mg/l (± 20 ppm)
- Use of the chloride electrode for compensation:
For concentrations > 500 mg/l (> 500 ppm) with simultaneously fluctuating values of ± 100 mg/l (± 100 ppm)

Manual offset

Constant systematic measured errors, which occur over the entire concentration range, can be corrected by setting a suitable manual offset. Here, the set offset is added to the measured value. To correct the measured error, a suitable negative value (often ranging from - 0.2 to 2 mg/l (- 0.2 to 2 ppm) for municipal wastewater) must be set as the manual offset for the specific ion-selective electrodes.

Using the offset for potassium or chloride values that do not fluctuate.

When determining ammonium, a manual offset of -1 mg/l $\text{NH}_4\text{-N}$ (-1 ppm $\text{NH}_4\text{-N}$) should be set per 20 mg/l (20 ppm) of potassium for complete compensation. When determining

nitrate, the manual offset should be - 1 mg/l NO₃-N (-1 ppm NO₃-N) per 200 mg/l (200 ppm) chloride. It is normally not necessary to set a manual offset when using ion-selective electrodes for potassium and chloride, as the effect of interference ions on the measured value for potassium or chloride is too small. The offset value can be left at zero.

Checking the calibration

1. Take 3 liters (0.79 US gal.) of sample from the wastewater treatment plant outlet.
2. Provide a bucket with drinking water.
3. Transfer exactly 2 liters (0.53 US gal.) of the sample into a suitable vessel.
4. Immerse the sensor into the sample.
5. Make sure there is convection in the solution (use a magnetic stirrer with a magnetic stir bar or constantly move the sensor slightly by hand).
 - ↳ After a few minutes the measured value should match the reference measured value (laboratory value) within the normal tolerance for measured value fluctuations.
6. Have some of the sample analyzed in the laboratory for the parameter to be calibrated.
7. Gradually increase the concentration of the ion to be measured in the sample. Preferably use a microliter pipette to add defined volumes to the standard solution.
8. After waiting 5 to 10 minutes, note down the stable measured value.
 - ↳ The increase in the measured value should be as expected. The increase in concentration is calculated using the following formula: concentration increase = volume added x standard concentration x molar mass of parameter / (volume presented + total volume added).
9. Immerse the sensor in a bucket with drinking water.
10. Check the concentration and the raw values.
 - ↳ Typically, the values for ammonium are close to 0 mg/l for raw values of -170 mV or less. With 3 mg/l of nitrate, a raw value of at least +150 mV or higher should be achieved.

Example

In 5 individual steps, 0.5 ml of 1M ammonium nitrate standard solution are added each time to 2 liters of sample solution. The molar mass of NH₄-N and NO₃-N is 14 g/mol in each case. As the volume added is so small, the increase in the volume of sample solution can be disregarded. Each time standard solution is added, the concentration of NH₄-N and NO₃-N produced increases by
 $0.5 \text{ ml} * 1 \text{ mol/l} * 14 \text{ g/mol} / 2000 \text{ ml} = 3.5 \text{ mg/l (3.5 ppm)}$.

If the measured values do not increase as expected or are systematically too high or too low, implement the measures shown in the table.

Problem	Reason	Action
Measured values are always too high by the same amount	Manual offset setting is not negative or not negative enough	► Make manual offset more negative.
Measured values are always too low by the same amount	Manual offset setting is too negative	► Reduce the manual offset setting based on the amount.
Measured values are always too high by a certain percentage	Zero point setting is too low	► Calibrate the zero point.
Measured values are always too low by a certain percentage	Zero point setting is too high	

Problem	Reason	Action
Measured values are too high when concentrations are low and too low when concentrations are high	Manual offset setting is not negative enough and zero point set too high	► Make manual offset more negative, and repeat the calibration (preferably sample calibration or standard addition).
Measured values are too low when concentrations are low and too high when concentrations are high	Manual offset setting is too negative and zero point set too low	► Reduce the manual offset setting based on the amount, and repeat the calibration (preferably sample calibration or standard addition).
Nonlinear activation, average measured values too high	Slope setting is too high	► Calibrate the slope and zero point (preferably using standard addition with at least 2 volumes of standard added).
Nonlinear activation, average measured values too low	Slope setting is too low	

8 Diagnostics and troubleshooting

When troubleshooting, consider the entire measuring point:

- Transmitter
- Electrical connections and cables
- Assembly
- Sensor

The possible causes of error in the following table relate mainly to the sensor.

Problem	Testing	Solution
No display, no sensor reaction	Line voltage at transmitter?	► Connect mains voltage.
	Sensor connected correctly?	► Establish correct connection.
	Medium flow present?	► Create medium flow.
	Formation of build-up	► Clean sensor.
Display value too high or too low	Air bubbles present?	► Eliminate air bubbles by tapping on the sensor shaft.
	Sensor calibrated?	► Calibrate.
Display value fluctuating greatly	Air bubbles present?	► Eliminate air bubbles by tapping on the sensor shaft.
	Check mounting location.	► Select a different mounting location.
Display value is always in the range of 0 ± 15 mV	Moisture on the electrode plug-in head	► Remove moisture. ► Replace the electrode if necessary.
	Has the membrane cap been tightened by hand?	► Verify that the membrane cap is hand-tight.



Pay attention to the troubleshooting information in the Operating Instructions for the transmitter. Check the transmitter if necessary.

9 Maintenance

Take all the necessary precautions in time to ensure the operational safety and reliability of the entire measuring system.

NOTICE

Effects on process and process control!

- ▶ When carrying out any work on the system, bear in mind any potential impact this could have on the process control system and the process itself.
- ▶ For your own safety, only use genuine accessories. With genuine parts, the function, accuracy and reliability are also ensured after maintenance work.

9.1 Maintenance schedule

NOTICE

Moisture on the electrode contacts

Results in short-circuits and consequently in drifting or unstable measured values

- ▶ When working with ion-sensitive electrodes, verify that the contacts are dry.
- ▶ Do not touch the plug contacts using bare hands.

	Cleaning	Membrane cap and electrolyte replacement			Crystal polishing	Replacing	
Maintenance interval	Membrane	Ammonium	Nitrate	Potassium	Chloride	pH electrode	O-rings
Fortnightly	☑						
Biannually		☑	☑	☑	☑		
Annually						☑	☑



The intervals indicated are average empirical values and can be shorter or longer depending on the operating conditions. You, or the plant operator, are responsible for adapting the intervals to suit your conditions.

9.2 Cleaning the membrane

If the membrane is very contaminated, clean it regardless of the maintenance intervals.

- Do not touch the membrane with your hands.
- Use a clean tissue and water for cleaning purposes.

The optional chloride electrode has a crystal instead of a membrane. To clean, proceed as follows:

1. Place some sandpaper (600-grit) on an even surface.
2. With the crystal area facing downwards, rub the sensor on the paper until all residues of dirt are removed.
3. Perform a visual inspection. Rubbing the sensor for a few seconds usually suffices.

9.3 Replacing membrane cap and electrolyte

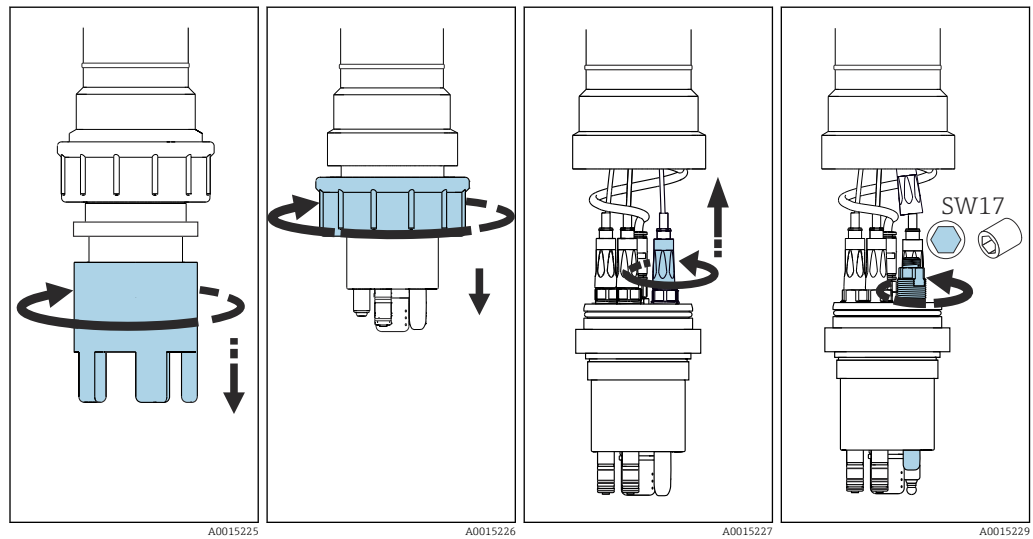
NOTICE

Sensor out of medium for longer than 15 minutes and did not wait for conditioning

Results in measurement errors

- ▶ After submerging the sensor in the medium, you must allow it time for conditioning. Allow approx. 12 hours for this.

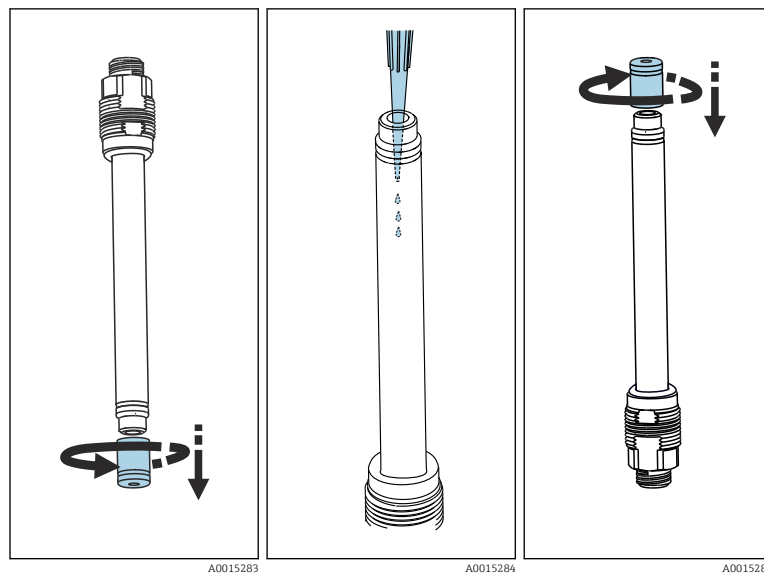
Removing the electrode



- 10 Release protection guard
 11 Remove coupling nut
 12 Release cable
 13 Remove electrode





1. Remove the sensor from the medium.
2. Clean the sensor with water.
3. Release the protection guard and remove (→ 10, 24).
4. Unscrew the coupling nut (→ 11, 24).
5. Pull the electrode holder out of the sensor and release the electrode cable of the electrode to be replaced (→ 12, 24).
6. Remove the electrode using a socket wrench, AF17 → 13, 24.


Replacing the membrane cap and electrolyte












- 14 Remove cap
 15 Fill with electrolyte
 16 New cap

1. Unscrew the membrane cap from the electrode (→ 10, 24).
2. Dispose of the membrane cap as waste.
3. Drain electrolyte from electrode body.
4. Take fresh electrolyte from the supply bottle using the pipette supplied in the kit.

5. Fill the electrode body with electrolyte up to approx. 2-3 mm (0.08 - 0.12") under the rim (→  11,  24).
6. Dry the thread of the electrode carefully.
7. Continue to hold the electrode upright, with the cable connection head facing downwards.
8. Screw on the membrane cap until it is hand-tight (→  16,  24).
9. Turn the electrode around.
10. Eliminate any air bubbles on the inner membrane surface by holding the electrode vertically and shaking it vigorously several times (as in the case of a clinical thermometer).

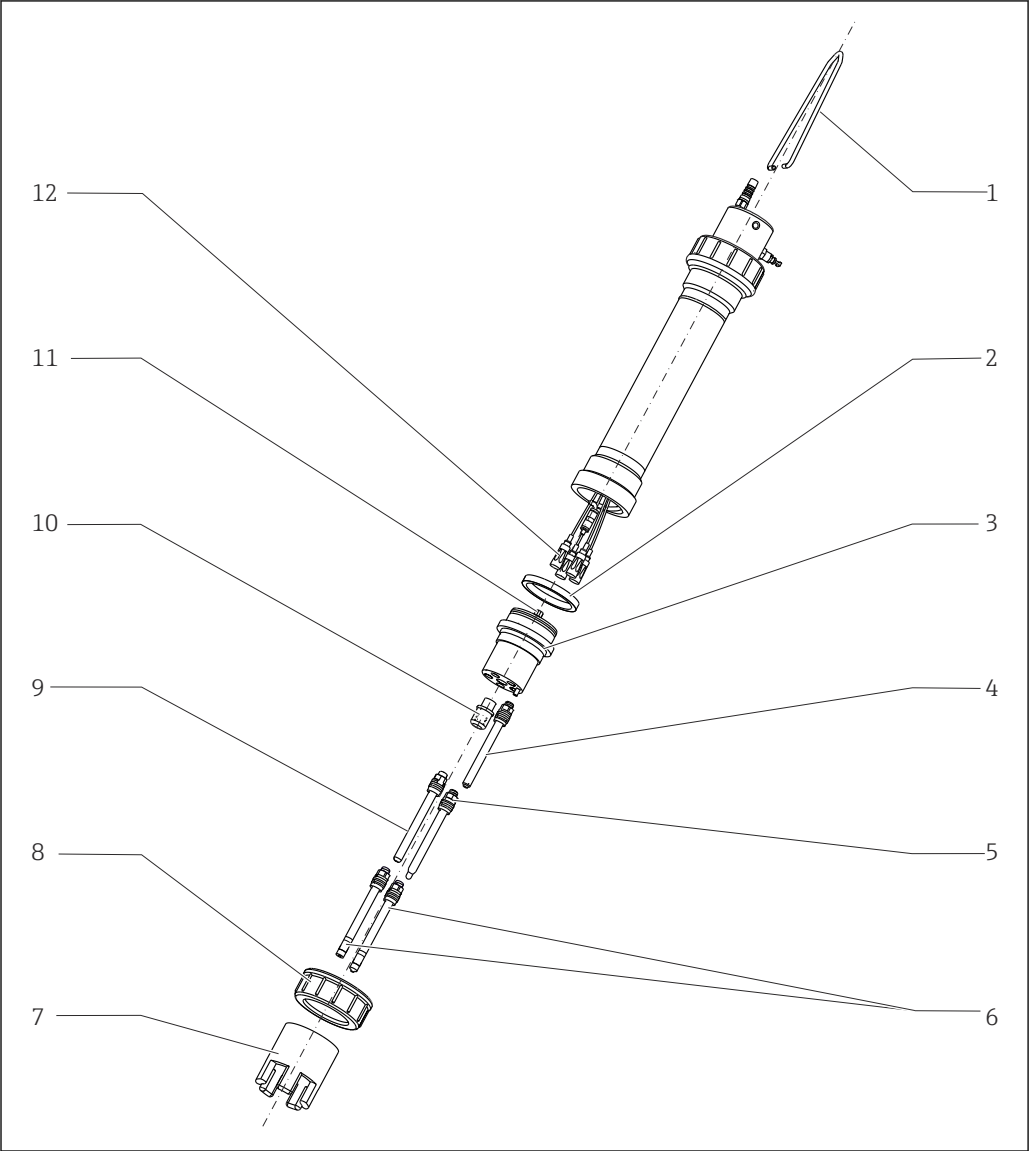
 From now until installation in the process, always hold the electrode and sensor upright to avoid the accumulation of new air bubbles on the inner membrane surface.

Installing the electrode

1. Screw the electrode into the electrode holder.
2. Tighten hand-tight with the socket wrench (→  13,  24, but in the opposite direction).
3. Connect the electrode connector to the cable (→  12,  24, opposite direction).
4. Carefully push the electrode holder and the air hose back into the sensor.
5. Screw on the coupling nut (→  11,  24, opposite direction). When doing so, pay attention to the radial seal on the electrode holder, and apply more grease if necessary.
6. Screw on the protection guard (→  10,  24, opposite direction).
7. Perform a calibration (→  16).

10 Repair

10.1 Spare parts



A0015217

17 Spare parts CAS40D

No.	Designation	Order no.
1	Kit CYH112 suspension bracket for chain	71096714
2	Kit CAS40D seal set <ul style="list-style-type: none">■ Silicone grease, 2 g■ 2 x O-ring ID 69.44 mm, width 3.53 mm■ 5 x O-ring ID 11 mm, width 2.50 mm■ O-ring ID 18 mm, width 4 mm■ Kit manual	71260474
3, 10, 11	Kit CAS40D electrode holder <ul style="list-style-type: none">■ Electrode holder■ Seals for electrodes■ Radial seal for electrode holder (3)■ Cleaning nozzle (10) incl. seal■ Check valve (11)	71260473

No.	Designation	Order no.
4	Temperature sensor	CTS1-A2GSA
5	pH sensor with reference	CPS11-1AS2GSA
6	Ion-selective electrodes, electrode complete, length 120 mm <ul style="list-style-type: none"> ■ Ammonium ■ Nitrate ■ Potassium ■ Chloride 	71109938 71109937 71109936 71109939
7	Kit CAS40D electrode protection guard	71130354
9	Kit CAS40D dummy electrode (needed to seal off unused slots)	71123812
10	Kit CAS40D cleaning nozzle incl. seal	71130359
12	Kit CAS40D multi-conductor cable for electrodes	71130358

10.2 Return

The product must be returned if repairs or a factory calibration are required, or if the wrong product was ordered or delivered. As an ISO-certified company and also due to legal regulations, Endress+Hauser is obliged to follow certain procedures when handling any returned products that have been in contact with medium.

To ensure the swift, safe and professional return of the device:

- Refer to the website www.endress.com/support/return-material for information on the procedure and conditions for returning devices.

10.3 Disposal

The device contains electronic components. The product must be disposed of as electronic waste.

- Observe the local regulations.

11 Accessories

The following are the most important accessories available at the time this documentation was issued.

- For accessories not listed here, please contact your Service or Sales Center.

11.1 Assembly holder

Flexdip CYH112

- Modular holder system for sensors and assemblies in open basins, channels and tanks
- For Flexdip CYA112 water and wastewater assemblies
- Can be affixed anywhere: on the ground, on the coping stone, on the wall or directly onto railings.
- Stainless steel version
- Product Configurator on the product page: www.endress.com/cyh112



Technical Information TI00430C

11.2 Maintenance kits

Membrane kit

- 2 membrane caps (apart from chloride, where there is only one cap, with crystal)
- Electrolyte
- Order numbers:
 - Ammonium: 71072574
 - Nitrate: 71072575
 - Potassium: 71072576
 - Chloride: 71072577

Maintenance kit for chloride electrode

- Sandpaper
- Electrolyte
- Order number: 71085727

11.3 Electrodes

Ion-selective electrode

- Electrode, complete, length 120 mm
- Order numbers:
 - Ammonium: 71109938 (color identification red)
 - Nitrate: 71109937 (color identification blue)
 - Potassium: 71109936 (color identification yellow)
 - Chloride: 71109939 (color identification green)

pH electrode with reference

Order number: CPS11-1AS2GSA

Temperature sensor

Order number: CTS1-A2GSA

Dummy electrode

Order number: 71123812

11.4 Standard solutions

CAY40

- Standard solutions for ammonium, nitrate, potassium and chloride
- Ordering information: www.endress.com/cas40d under "Accessories/spare parts"

High-quality buffer solutions from Endress+Hauser - CPY20

The secondary buffer solutions have been referenced to primary reference material of the PTB (German Federal Physico-technical Institute) or to standard reference material of NIST (National Institute of Standards and Technology) according to DIN 19266 by a laboratory accredited by the DAkkS (German accreditation body) according to DIN 17025. Product Configurator on the product page: www.endress.com/cpy20

11.5 Compressed air cleaning

Not suitable for continuous operation!

- Operating interval: max. 3 minutes cleaning, break for at least six times the cleaning time.
- Avoid condensation in the pressurized hoses.

Cleaning unit in the housing

- 230 V or 115V, IP 65
- Conveying rate at atmospheric pressure: 50 l/min (13.2 gal/min)
- Power consumption: 240 W
- Current consumption: 1.3 A
- Overheating protection: automatic switch off at $T > 130\text{ °C}$ (266 °F)
- Order no.
 - 230 V: 71072583
 - 115 V: 71194623
 - Hose reducer coupling AD 8/6 mm: 71082499

12 Technical data

12.1 Input

Measured values	Depending on version: <ul style="list-style-type: none"> ■ Ammonium: $\text{NH}_4\text{-N}$, NH_4^+ [mg/l] ■ Nitrate: $\text{NO}_3\text{-N}$, NO_3^- [mg/l] ■ Potassium, K^+ [mg/l] ■ Chloride, Cl^- [mg/l] ■ pH value ■ Temperature
Measuring ranges	<ul style="list-style-type: none"> ■ Ammonium: 0.1 to 1000 mg/l ($\text{NH}_4\text{-N}$) ■ Nitrate: 0.1 to 1000 mg/l ($\text{NO}_3\text{-N}$) ■ Potassium: 1 to 1000 mg/l ■ Chloride: 1 to 1000 mg/l

12.2 Performance characteristics

Response time t_{90} of the ion-selective sensors	< 2 min. For a change between 0.5 and 1 mmol/l in both directions, at 25 °C (77 °F).
Measured error	$\pm 5\%$ of the measured value ± 0.2 mg/l
Repeatability	$\pm 3\%$ of the display value

Compensation	Sensor	Temperature	pH	Potassium ^{1) 2)}	Chloride ^{3) 4)}
	Ammonium	2 to 40 °C (36 to 100 °F)	pH 8.3 to 10	1 to 1000 mg/l (ppm)	-
	Nitrate		-	-	10 to 1000 mg/l (ppm)
	Potassium		-	-	-
	Chloride		-	-	-

- 1) The concentration fluctuations, not the absolute value, are decisive
- 2) Recommendation: Use as compensation electrode for potassium concentrations > 40 mg/l in the case of simultaneously fluctuating values of ± 20 mg/l, or apply an offset in the case of non-fluctuating values.
- 3) The concentration fluctuations, not the absolute value, are decisive
- 4) Recommendation: Use as compensation electrode for chloride concentrations > 500 mg/l in the case of simultaneously fluctuating values of ± 100 mg/l, or apply an offset in the case of non-fluctuating values.

Max. operating life	Membrane and electrolyte <ul style="list-style-type: none"> ■ Use: approx. 0.5 years ■ Storage: 2 years
---------------------	---

Automatic cleaning	<ul style="list-style-type: none"> ■ Cleaning medium: Air ■ Pressure: 3 to 3.5 bar (45 to 50 psi) ■ Volume of air required per cleaning cycle: 3 to 4 l (0.8 to 1 US gal) ■ Cleaning duration: 4 to 15 s ■ Cleaning intervals (at T > 10 °C (50 °F)): Sludge activation inlet: 15 s cleaning, 30 min pause Sludge activation: 15 s cleaning, 1 hr pause
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12.3 Environment

Ambient temperature	−20 to 50 °C (−4 to 122 °F)
Storage temperature	2 to 40 °C (36 to 104 °F)
Degree of protection	IP68 (2 m water column, 25 °C, 48 h)
Electromagnetic compatibility	Interference emission and interference immunity as per EN 61 326, Namur NE21

12.4 Process

Process temperature	2 to 40 °C (36 to 104 °F)
Process pressure	400 mbar (160 in H ₂ O) max. permitted overpressure
pH value of the medium	<ul style="list-style-type: none"> ■ Ammonium: pH 5 to 8.3 (without pH compensation) pH 5 to 10 (with pH compensation) ■ Nitrate: pH 2 to 12 ■ Potassium: pH 2 to 12 ■ Chloride: pH 1 to 10

12.5 Mechanical construction

Design, dimensions	→  9
Weight	Approx. 3.5 kg (7.7 lbs)

Materials

Sensor:

Protective cage:	POM
Electrode holder:	POM
Radial seal for sensor head and electrode holder:	Silicone
O-rings in ISE holder:	EPDM
O-rings for air nozzle:	VITON
Sensor pipe with coupling nut:	PP
Retaining bracket:	Stainless steel
Sensor head:	POM
Temperature sensor:	Glass
pH single-rod measuring cell with reference electrode:	Glass, PTFE

Ion-selective electrodes

Membrane cap:	POM
Shaft:	POM
Color ring:	PP
Membrane:	PVC, plasticizer
O-rings:	EPDM

Materials not in contact with the medium

The following specifications refer to the built-in temperature sensor CTS1.

Information according to REACH Regulation (EC) 1907/2006 Art. 33/1:

The potting compound in the sensor shaft contains the SVHC substance terphenyl, hydrogenated (CAS number ¹⁾ 61788-32-7) with more than 0.1 % (w/w). The product does not present a hazard if it is used as designated.

Electrode process
connection

Pg 13.5

Compressed air connection

For hose, OD 8 mm

1) CAS = Chemical Abstracts Service, international identification standard for chemical substances

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